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Written by Gary Miessler, Paul Fischer, and Donald Tarr, this manual includes fully worked-out solutions to all end-of-chapter problems in the text. This product accompanies Inorganic Chemistry, 5th Edition

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The lobes of the three p-orbitals in atom A are present along the axes..., and orbitals have lobes along the axis. Hence, the symmetry of and orbitals of atom B can match with the symmetry of p-orbitals of atom A for end-to-end overlapping,. Therefore, form sigma bonds can be formed between these orbitals.

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