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- 2 - Section A Answer all questions. Write your answers in the boxes provided. 1. Ethanedioic acid is a diprotic acid. A student determined the value of x in the formula of hydrated ethanedioic acid, $\text{HOOC-COOH}\cdot x\text{H}_2\text{O}$, by titrating a known mass of the acid with a 0.100 mol dm⁻³ solution of NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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That chemistry exam was horrible, I was aiming for a 6/7 and I've gotten that in every past paper I've done from 2016&2017 and now I'm praying for a 3. That paper 2 was what the fuck, it was so many formulas I've never used before and it was a hell lot of physics?

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